



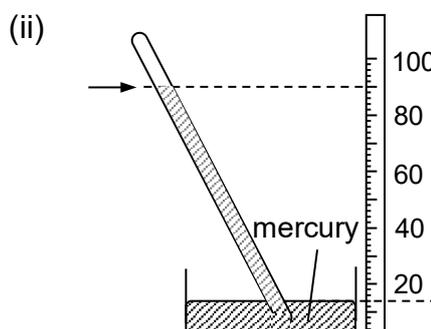
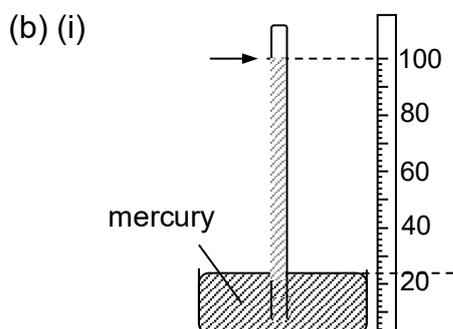
2025 Sec 4 Physics Pressure Assignment 9 A09 Answers

Note:

- Always state the formula or equation used!
- Show all working steps in calculation or reasoning.

1 (a) (i) $(90 \text{ cm} - 14 \text{ cm}) \text{ Hg} = \mathbf{76 \text{ cm Hg}}$

(ii) $(90 \text{ cm} - 40 \text{ cm}) \text{ Hg} = \mathbf{50 \text{ cm Hg}}$



2 (a) $P = 75.0 \text{ cm Hg}$

$$P = h\rho g = 0.750 \text{ m} \times 13600 \text{ kg m}^{-3} \times 10 \text{ N kg}^{-1} = 1.02 \times 10^5 \text{ Pa}$$

Note:

- tube P is like a normal mercury barometer, the space at upper end is a vacuum ($P = 0$)
- tube Q is like a faulty barometer, with air trapped at the upper end space (P_{air})
- Consider two points **A** & **B** at the same level of the mercury in the container which are at the same pressure.

(b)

$$P_A = P_B$$

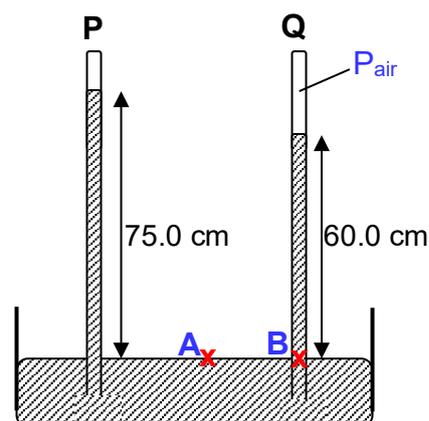
$$75.0 \text{ cm Hg} = P_{\text{air}} + 60.0 \text{ cm Hg}$$

$$P_{\text{air}} = 15.0 \text{ cm Hg}$$

$$P_{\text{air}} = h\rho_{\text{Hg}}g$$

$$= 0.150 \text{ m} \times 13600 \text{ kg m}^{-3} \times 10 \text{ N kg}^{-1}$$

$$= 2.04 \times 10^4 \text{ Pa}$$



3

(a)

$$P_C = P_D$$

$$P_{\text{gas}} = P_{\text{atm}} + (40 - 21) \text{ cm Hg}$$

$$= 75 \text{ cm Hg} + (40 - 21) \text{ cm Hg} = 94 \text{ cm Hg}$$

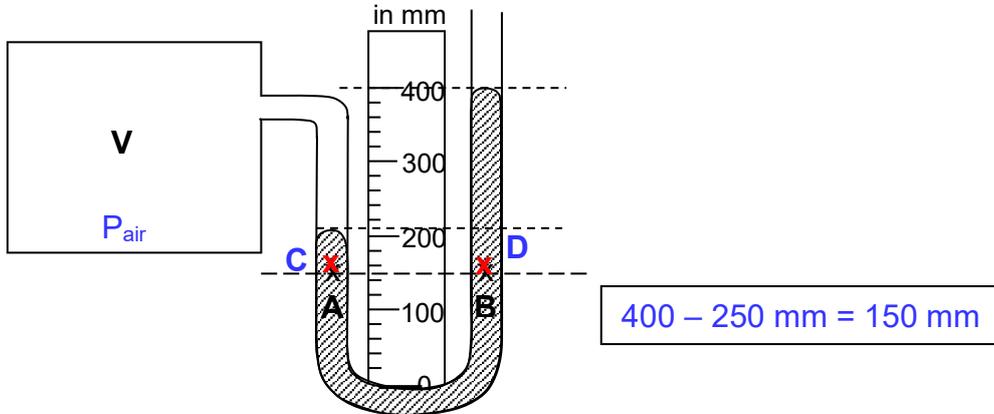
Note:

- Consider two points **C** & **D** at the same level of the mercury in the manometer which are at the same pressure (choose the lower of the 2 levels).

Note:

- Consider two points **C** & **D** (label these) at the same level of the mercury in the manometer which are at the same pressure (choose the lower of the 2 levels).

3
(b)



4 (a) $P_A = 76 \text{ cm Hg}$

Note:

- Consider a point **C** (label this point) at the same level of the mercury as point **B** so both points are at the same pressure.

(b) $P_B = 76 + 22 = 98 \text{ cm Hg}$

(c) Left tube level: $(42 + 5) \text{ cm mark} = 47 \text{ cm mark}$

Right tube level: $(20 - 5) \text{ cm mark} = 15 \text{ cm mark}$

(d) (i) **32 cm mark**

- (ii) Since the gas pressure remains constant, the excess pressure remains constant i.e. 22 cm Hg. The height difference between the left and right arm remains at 22 cm.

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Pressure due to column of air of height h = Pressure difference measured at top and base of mountain (using mercury barometer)

$$\begin{aligned} H \rho_{air} g &= h_{bottom} \rho_{Hg} g - h_{top} \rho_{Hg} g \\ &= (h_{bottom} - h_{top}) \rho_{Hg} g \\ &= (0.750 - 0.580) \text{ m} \times 13600 \text{ kg m}^{-3} \times 10 \text{ N kg}^{-1} \end{aligned}$$

$$\begin{aligned} H \times 1.2 \times 10 &= 23120 \text{ Pa} \\ H &= 1930 \text{ m (to 3 s.f.)} \end{aligned}$$